

Synthesis and CD Spectra of Fluoro- and Hydroxy-Substituted β -Peptides

Preliminary Communication

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β -Amino acids **1–3** with OH and F substituents in the α -position have been prepared (*Scheme*) from the natural (*S*)- α -amino acids alanine, valine, and leucine, and incorporated into β -hexa- and β -heptapeptides **4–12**. The peptide syntheses were performed according to a conventional solution strategy (Boc/Bn protection) with fragment coupling. The new β -peptides with (series **a**) and without (series **b**) terminal protection were isolated in HPLC-pure form and characterized by NMR spectroscopy and MALDI mass spectrometry. The chemical properties as well as the patterns of the CD spectra (*Figs. 3–5*) depend upon constitution (OH, F, F₂ substitution) and configuration (*l* or *u*) of the amino acid residues, upon the total number of OH and F substituents in the peptide chain, and upon the solvent used (H₂O, MeOH, CF₃CH₂OH, (CF₃)₂CHOH). No reliable clues regarding the structures can be obtained from these CD spectra. Only a full NMR analysis will be able to answer the questions: *a*) with which known secondary structures (*Figs. 1* and *2*) of β -peptides are the OH and F derivatives compatible? *b*) Are new secondary structures enforced by the polar and/or H-bonding backbone substituents? Furthermore, the β -peptides described here will enable us to study changes in chemical, enzymatic, and metabolic stability, and in physiological properties caused by the heteroatoms.

Structural evidence from NMR, X-ray, molecular-modelling, and CD investigations of β -peptides containing the side chains of proteinogenic amino acids has provided rules for the design of the secondary structures 3_{14} helix and strands [1], while the possible substitution patterns and configurations of amino acids in the β -peptidic hairpin turn and *12/10* helix have not been probed as well [2]. As outlined in *Fig. 1*, steric hindrance should prevent non-H-atoms from occupying axial positions in the 3_{14} helix and in-plane position in a sheet (see black spheres in *Fig. 1,a* and *b*), which has been concluded mainly from investigations with Me substituents (**A**, **B**, and **C** in *Fig. 2*). The *Van der Waals* radius of a Me group is of course much larger than that of a H-atom, and we wondered whether an F- or O-atom, or an OH group might fit the forbidden position next to the C=O group (*cf.* the A values in *Fig. 1,c*, and **D**, **E**, **F**, and **G** in *Fig. 2*). As compared to the H-atom, these atoms have, of course, not only a somewhat different size but also an electronic effect (*Pauling* electronegativities H 2.2, C 2.6, O 3.4, F 4.0) and the ability of forming H-bonds⁴⁾. Thus, another issue to be addressed by the present study was to see what influence the substituents F and OH might exert on

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4) For a discussion of the H-bond-acceptor properties of organic fluoride, see [3].

the backbone conformation of a β -peptide when placed in allowed positions (green spheres in *Fig. 1*), an effect that cannot possibly be studied in α -peptides⁵⁾6). Finally, the F- and OH-substituted β -peptides will offer an opportunity to determine what effects the enhanced electrophilicity of the amide C=O groups has on their biological properties (such as antimicrobial and haemolytic activity, enzymatic and metabolic stability, or biodegradability⁷⁾).

The required enantiomerically pure *N*-Boc-protected 3-amino-2-fluoro and 3-amino-2-hydroxy carboxylic acids with the side chains of alanine, valine, and leucine were prepared as shown in the *Scheme*. The *N,N*-dibenzyl-protected aldehydes prepared from the amino acids by standard procedures [8] were subjected to a *trans*-cyanohydrination process (cyanohydrine of acetone/KCN/cat. Bu₄NI [9]) to give mixtures (1:2 to 1:1) of epimeric cyanohydrins. This nonselective reaction was practical in the present investigation, since we needed to have both, the *l*- and the *u*-diastereoisomers of the 3-amino-2-hydroxyalkanoates, and since it did not make any difference which epimer was used for the conversion to the difluoro derivatives. Methanolysis of the epimeric dibenzylamino-cyanohydrins, followed by chromatographic separation of the hydroxy methyl esters, debenzylation, *N*-Boc protection, and saponification, led to the configurationally pure acids **1a–1c**. On the other hand, the diastereoisomeric *N,N*-dibenzyl-protected hydroxy methyl esters were converted to the 2-fluoromethyl esters by retentive OH/F substitution with DAST [10][11], saponified, debenzylated, and *N*-Boc-protected to give the acids **2a–2c** of either (*R,S*)- or (*S,S*)-configuration. Oxidation of the epimeric mixtures of 2-hydroxy methyl esters to 2-keto esters, which do not racemize if chromatographic purification is avoided, treatment with DAST [10], saponification, debenzylation, and *N*-Boc protection provided the 2,2-difluoro carboxylic acids **3a–3c** (*Scheme*). For the C-terminal amino acid residues of the β -peptides to be prepared (see **4–13**) by solution coupling of *N*-Boc-protected hydroxy and fluoro amino acids, the benzyl esters of **1c**, **2c**, and **3c** were required; they were prepared by treatment of the corresponding Cs salts with BnBr [12].

With the suitable hydroxy and fluoro amino acids at hand, we synthesized the β -tri-, β -hexa-, and β -heptapeptides **4–13** with various numbers of hydroxy or fluoro amino acids incorporated. All syntheses were carried out in solution and according to the same principle: starting from the benzyl ester of the C-terminal Leu-derived hydroxy or fluoro amino acid, a tripeptide was built up⁸⁾, which was then employed for a dimerizing fragment coupling to an *N*- and *C*-protected hexapeptide **6a**, **10a**, **11a**, and **12a**. When a central OH-, F-, or F₂-substituted β -amino acid was to be incorporated, the β -tripeptide with the side chains of Val, Ala, and Leu was first coupled with the corresponding heterosubstituted β -amino acid, and, to the resulting tetrapeptide, another β^3 hVal- β^3 hAla- β^3 hLeu fragment was attached to give the β -heptapeptides **4a**, **5a**, **7a**, **8a**, and **9a**. Full deprotection (Bn and *N*-Boc cleavage) of β -peptides **4a–9a**,

5) For a general discussion of OH substitution in β - and γ -peptides, see [4].

6) We know already that a β -peptide consisting of *unlike*-3-amino-2-hydroxy carboxylic acid moieties (*cf.* **11**) folds to a novel 2₈ helix (probably due to H-bond formation of OH with the neighboring C=O O-atom [5]), rather than adopting a sheet structure [6].

7) For investigations of these properties of 'normal' β -peptides, see [7], and refs. cit. therein.

8) Boc- β^3 hVal- β^3 hAla- β^3 hLeu-OBn has been used by us many times before [1].

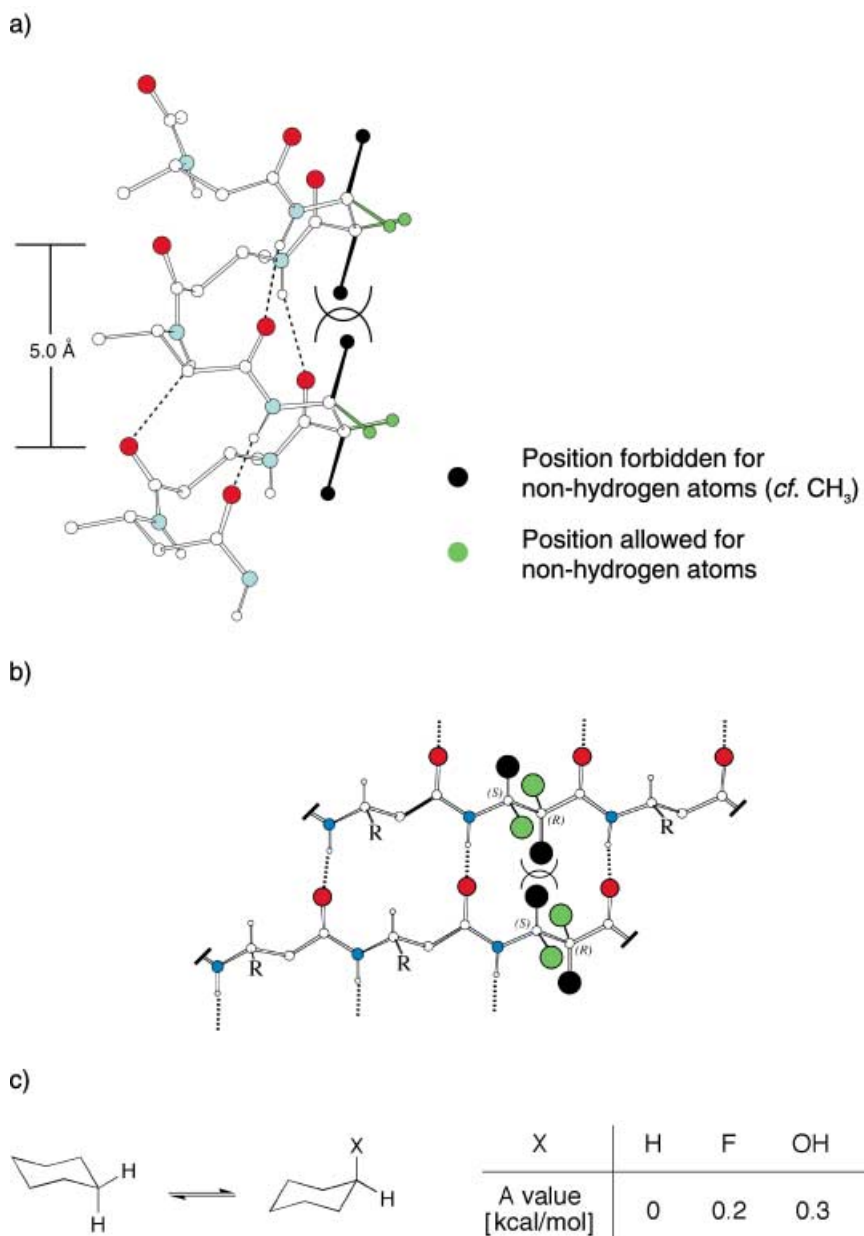
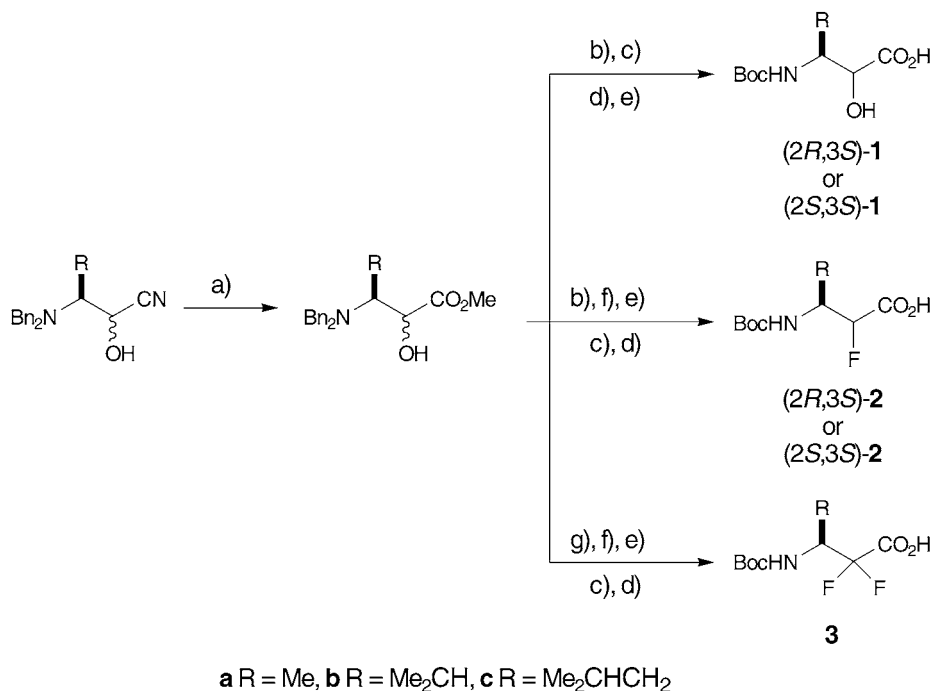


Fig. 1. Model of the (M)-3₁₄ helix of a β -peptide (a) and of the β -peptidic parallel sheet structure (b) and A values for preference of equatorial vs. axial position on cyclohexane (c)

followed by reversed-phase HPLC purification, provided the *N*- and *C*-terminally unprotected β -peptides **4b–9b**.

Scheme. Preparation of the *N*-Boc-Protected 3-Amino-2-fluoro, 3-Amino-2,2-difluoro, and 3-Amino-2-hydroxy Carboxylic Acids **1–3** from the Proteogenic Amino Acids Alanine, Valine, and Leucine. The OH/F substitution with DAST is carried out with the *N,N*-dibenzyl-protected, diastereomerically pure methyl 3-amino-2-hydroxy-alkanoates and takes place with retention of configuration (as established by X-ray analysis of one of the products and spectroscopic assignment by analogy). Boc = *t*-(Butoxy)carbonyl, DAST = Et₂NSF₃.



a) HCl (gas)/MeOH. b) Chromatographic separation of epimers (SiO₂; hexane/AcOEt). c) H₂/Pd-C, MeOH. d) Boc₂O, Et₃N, MeOH, 20°. e) LiOH · H₂O, EtOH. f) DAST, CH₂Cl₂, 0°. g) Swern oxidation: DMSO, SOCl₂, CH₂Cl₂, -78°.

All hexa- and heptapeptides, of which the formulae are shown herein, have been isolated in high purity (>97%) and characterized by CD, ¹³C- and ¹H-NMR spectroscopy, and by MALDI mass spectrometry. The starting materials (amino acid derivatives) and the intermediate di-, tri-, and tetrapeptides have been fully characterized by NMR, IR, MS, [α]_D, m.p., and elemental analysis⁹).

Some CD spectra¹⁰) in 0.2 mM MeOH of the unprotected β³-heptapeptides, with central F- and OH-substituted amino acid moieties, are shown in Fig. 3. They are most puzzling, to say the least, because all show the ₃₁₄-helix-typical pattern with a negative Cotton effect between 215 and 220 nm, zero crossing between 210 and 205 nm, and a positive Cotton effect near 200 nm – no matter whether the F or OH group would occupy a lateral (**4b** and **7b**) or axial position (**5b**, **8b**, and **9b**) on such a ₃₁₄ helix. As shown in Fig. 4, the pattern of the CD spectra of the *u*-mono-hydroxy and of the *u*-

⁹) Full details will be presented in forthcoming full papers.

¹⁰) All CD spectra shown herein are non-normalized.

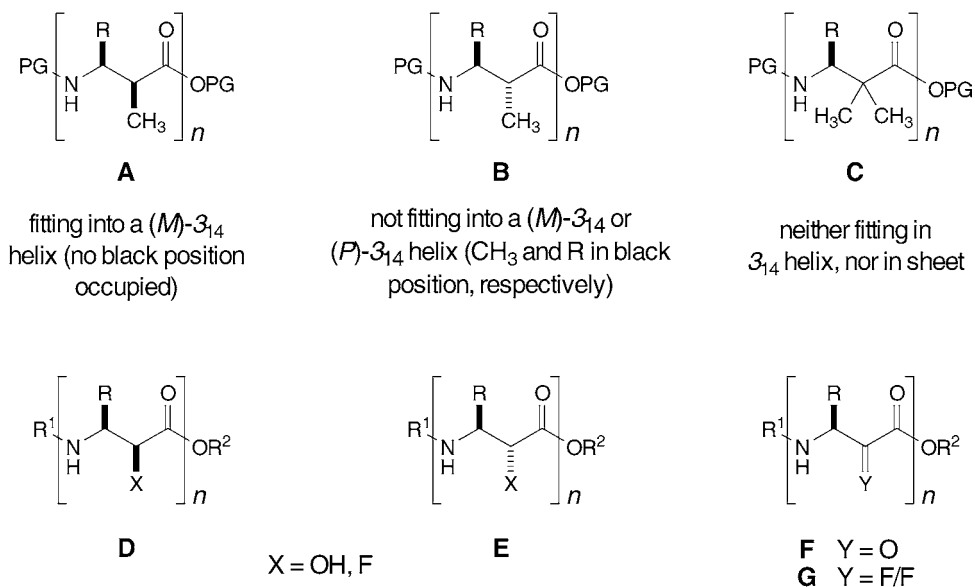


Fig. 2. General structures of $\beta^{2,3}$ - and $\beta^{2,2,3}$ -peptides **A–C** [1] and of the heterosubstituted β -peptides **D–G** described herein

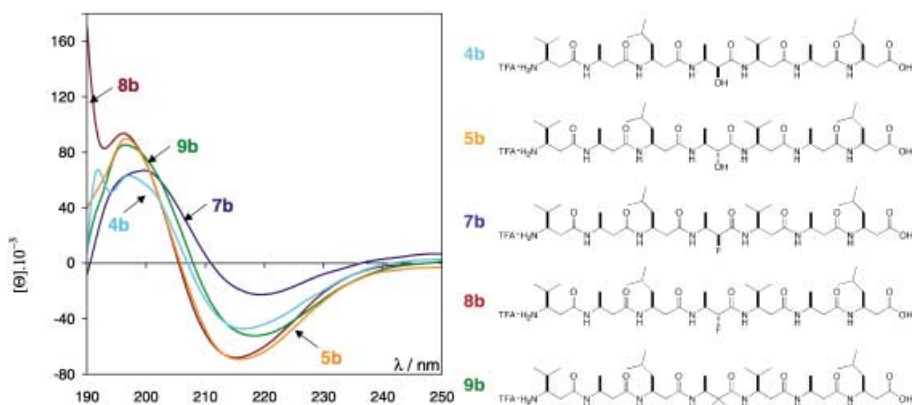
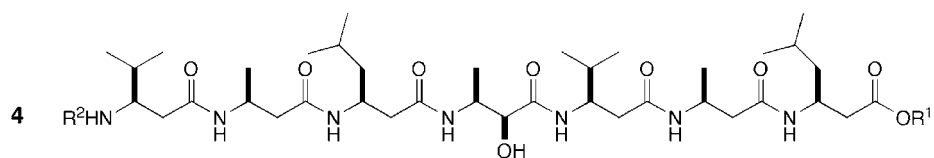


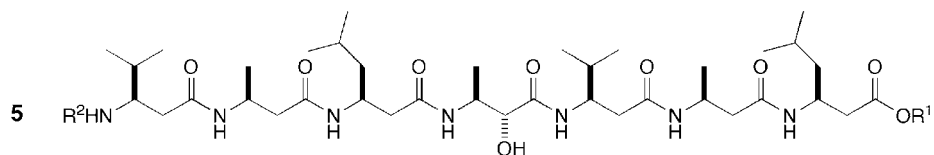
Fig. 3. Comparison of the CD spectra (0.2 mM in MeOH) of unprotected β -heptapeptides carrying a central α -heterosubstituted β -amino acid. All the new β -heptapeptides exhibit the CD pattern associated to an (*M*)- 3_{14} -helical structure [1].

mono-fluoro derivatives **5b** and **8b** does not change with the solvent (MeOH, $\text{CF}_3\text{CH}_2\text{OH}$, $(\text{CF}_3)_2\text{CHOH}$), except when we switch to aqueous solution, which causes a dramatic effect. Also, the protected and the unprotected¹¹⁾ β -peptides of this type may exhibit totally different CD spectra as exemplified by the mirror-image-type

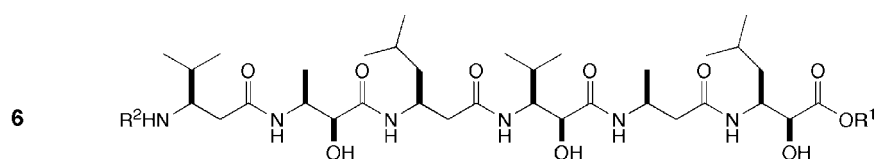
¹¹⁾ Cf. the stabilization of the 3_{14} helix upon terminal deprotection [1b][13].



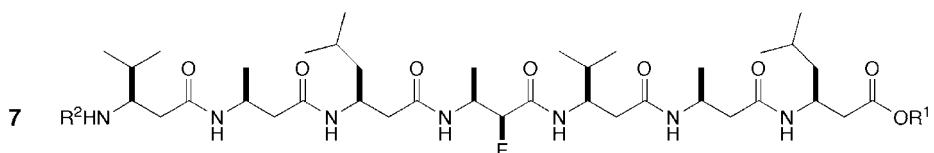
a R¹ = Bn, R² = Boc **b** R¹ = R² = H (TFA salt)



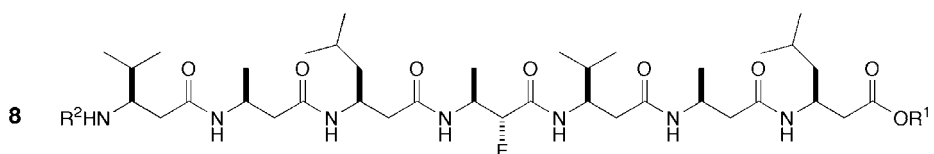
a R¹ = Bn, R² = Boc **b** R¹ = R² = H (TFA salt)



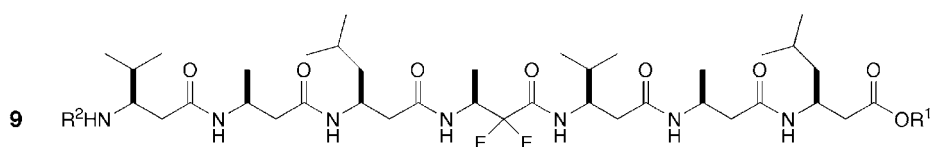
a R¹ = Bn, R² = Boc **b** R¹ = R² = H (TFA salt)



a R¹ = Bn, R² = Boc **b** R¹ = R² = H (TFA salt)



a R¹ = Bn, R² = Boc **b** R¹ = R² = H (TFA salt)



a R¹ = Bn, R² = Boc **b** R¹ = R² = H (TFA salt)

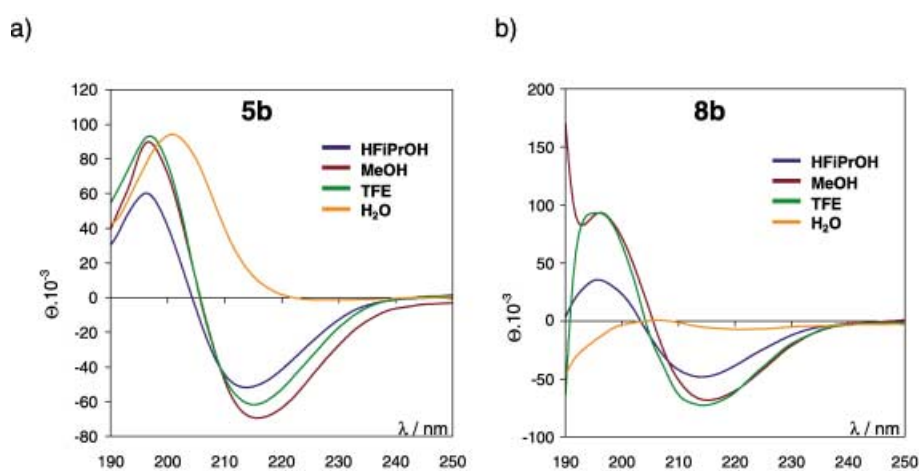
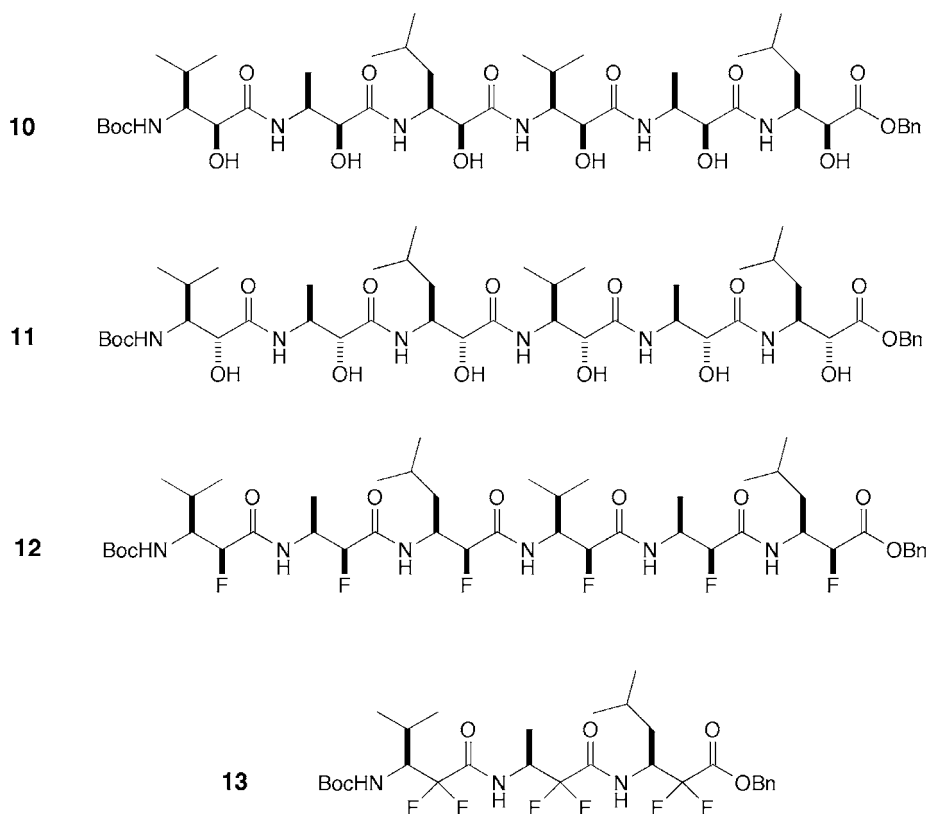


Fig. 4. Influence of the solvent on CD spectra (0.2 mM in MeOH, $\text{CF}_3\text{CH}_2\text{OH}$ (TFE), $(\text{CF}_3)_2\text{CHOH}$ (HFIPrOH), and H_2O) of the u-hydroxy- β -heptapeptide **5b** (a) and of the u-fluoro- β -heptapeptide **8b** (b).

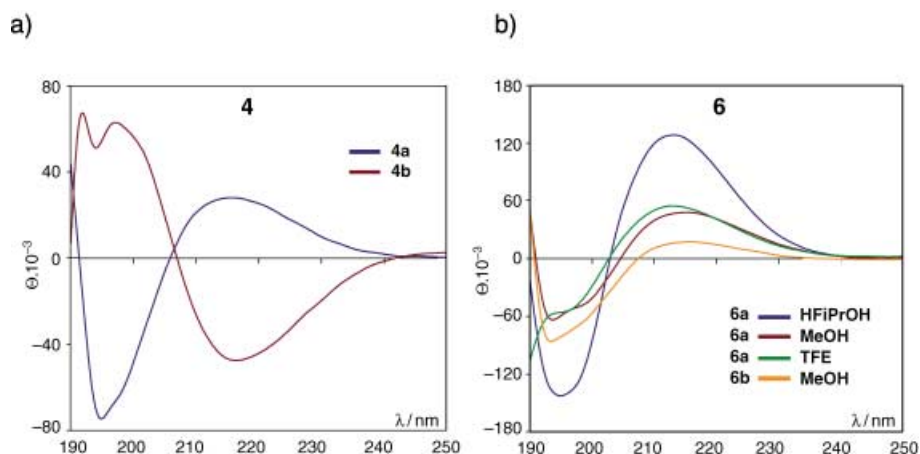


Fig. 5. CD Spectra of β -peptides **4** and **6** containing 1-2-hydroxy-3-amino acid moieties. a) CD Spectra in MeOH of the terminally protected and unprotected *l*-hydroxy- β -heptapeptides, **4a** and **4b**, respectively. b) Overlay of the CD spectra of the protected hexapeptide **6a** (in different solvents) and of its unprotected form **6b** (in MeOH).

curves obtained for the *l*-mono-hydroxy derivatives **4a** and **4b** in MeOH (Fig. 5,a). The all-*l*-trihydroxy-hexapeptide, on the other hand, shows CD spectra of the same general pattern, whether it is protected or unprotected, or whether MeOH or the fluorinated solvents are used (Fig. 5,b).

Thus, on the basis of the CD spectra, it is impossible to arrive at any safe structural assignment. If we would use analogy with the typical CD spectra of β -peptides, which, according to NMR analysis, form a 3_{14} helix [1][13], a $12/10$ helix [1][14], or a hairpin turn [6], we would, for instance, have to assign (*M*)- 3_{14} -helical structure to **4b**, **5b**, **7b**, **8b**, and **9b**, (*P*)- 3_{14} -helical structure to **4a** in MeOH and to **6a** in $(\text{CF}_3)_2\text{CHOH}$, and hairpin or $12/10$ -helical structure to **5b** in H_2O ! Thus, CD spectroscopy of β -peptides, once again, turns out to be useless for structure determination [15]. At this point, it can provide only a preliminary hint as to which compounds might be most interesting to investigate by NMR spectroscopy first! We have performed or are in the process of performing extensive, state-of-the-art, high-field NMR measurements of most of the novel β -peptides mentioned herein. The interpretation of these measurements will require time¹²⁾, and the results will be published in due course. There will be, no doubt, many surprises.

REFERENCES

- [1] a) D. Seebach, M. Overhand, F. N. M. Kühnle, B. Martinoni, L. Oberer, U. Hommel, H. Widmer, *Helv. Chim. Acta* **1996**, *79*, 913; b) D. Seebach, P. E. Ciceri, M. Overhand, B. Jaun, D. Rigo, L. Oberer, U. Hommel, R. Amstutz, H. Widmer, *Helv. Chim. Acta* **1996**, *79*, 2043; c) D. Seebach, S. Abele, K. Gademann,

¹²⁾ It usually takes *ca.* 3.5 days to perform the measurements (1D Presat, DQFCOSY, TOCSY, HSQC $^{13}\text{C}/^1\text{H}$, HMBC $^{13}\text{C}/^1\text{H}$, ROESY with different mixing times in ms), and it may take several weeks or even months for the evaluation of the data.

- G. Guichard, T. Hintermann, B. Jaun, J. L. Matthews, J. V. Schreiber, L. Oberer, U. Hommel and H. Widmer, *Helv. Chim. Acta* **1998**, *81*, 932; d) D. Seebach, T. Sifferlen, P. A. Mathieu, A. M. Häne, C. M. Krell, D. J. Bierbaum, S. Abele, *Helv. Chim. Acta* **2000**, *83*, 2849; e) D. Seebach, T. Sifferlen, D. J. Bierbaum, M. Rueping, B. Jaun, B. Schweizer, J. Schaefer, A. K. Mehta, R. D. O'Connor, B. H. Meier, M. Ernst, A. Glättli, *Helv. Chim. Acta* **2002**, *85*, 2877; and refs. cit. therein.
- [2] R. Baron, D. Bakowies, W. F. van Gunsteren, X. Daura, *Helv. Chim. Acta* **2002**, *85*, 3872; and refs. cit. therein.
- [3] J. D. Dunitz, R. Taylor, *Chem.–Eur. J.* **1997**, *3*, 89.
- [4] M. Brenner, D. Seebach, *Helv. Chim. Acta* **2001**, *84*, 1181.
- [5] A. Häne, K. Gademann, B. Jaun, D. Seebach, *Angew. Chem.* **2003**, *115*, 1573; *Angew. Chem., Int. Ed.* **2003**, *42*, 1534.
- [6] D. Seebach, S. Abele, K. Gademann, B. Jaun, *Angew. Chem.* **1999**, *111*, 1700; *Angew. Chem., Int. Ed.* **1999**, *38*, 1595.
- [7] a) J. Frackenhohl, P. I. Arvidsson, J. V. Schreiber, D. Seebach, *ChemBioChem* **2001**, *2*, 445; b) P. I. Arvidsson, J. Frackenhohl, N. S. Ryder, B. Liechty, F. Petersen, H. Zimmermann, G. P. Camenisch, R. Woessner, D. Seebach, *ChemBioChem* **2001**, *2*, 771; c) J. V. Schreiber, J. Frackenhohl, F. Moser, T. Fleischmann, H.-P. Kohler, D. Seebach, *ChemBioChem* **2002**, *3*, 424; d) H. Wiegand, B. Wirz, A. Schweitzer, G. P. Camenisch, M. I. Rodriguez Perez, G. Gross, R. Woessner, R. Voges, P. I. Arvidsson, J. Frackenhohl, D. Seebach, *Biopharm. Drug Dispos.* **2002**, *23*, 251.
- [8] a) J. M. Andrés, R. Barrio, M. A. Martinez, R. Pedrosa, A. Pérez-Encabo, *J. Org. Chem.* **1996**, *61*, 4210; b) M. T. Reetz, M. W. Drewes, R. Schwickardi, *Org. Synth.* **2000**, *76*, 110.
- [9] M. A. Schwindt, D. T. Belmont, M. Carlson, L. C. Franklin, V. S. Hendrikson, G. L. Karrick, R. W. Poe, D. M. Sobieray, J. Van De Vusse, *J. Org. Chem.* **1996**, *61*, 9564.
- [10] R. P. Singh, J. M. Shreeve, *Synthesis* **2002**, *17*, 2561.
- [11] a) L. Somekh, A. Shanzer, *J. Am. Chem. Soc.* **1982**, *104*, 5836; b) D. Gani, P. B. Hitchcock, D. W. Young, *J. Chem. Soc. Perkin Trans. 1* **1985**, 1363.
- [12] S.-S. Wang, B. F. Gisin, D. P. Winter, R. Makofske, I. D. Kulesha, C. Tzougraki, J. Meienhofer, *J. Org. Chem.* **1977**, *42*, 1286.
- [13] D. Seebach, J. V. Schreiber, S. Abele, X. Daura, W. F. van Gunsteren, *Helv. Chim. Acta* **2000**, *83*, 34.
- [14] D. Seebach, K. Gademann, J. V. Schreiber, J. L. Matthews, T. Hintermann, B. Jaun, L. Oberer, U. Hommel, H. Widmer, *Helv. Chim. Acta* **1997**, *80*, 2033.
- [15] A. Glättli, X. Daura, D. Seebach, W. F. van Gunsteren, *J. Am. Chem. Soc.* **2002**, *124*, 12972.

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